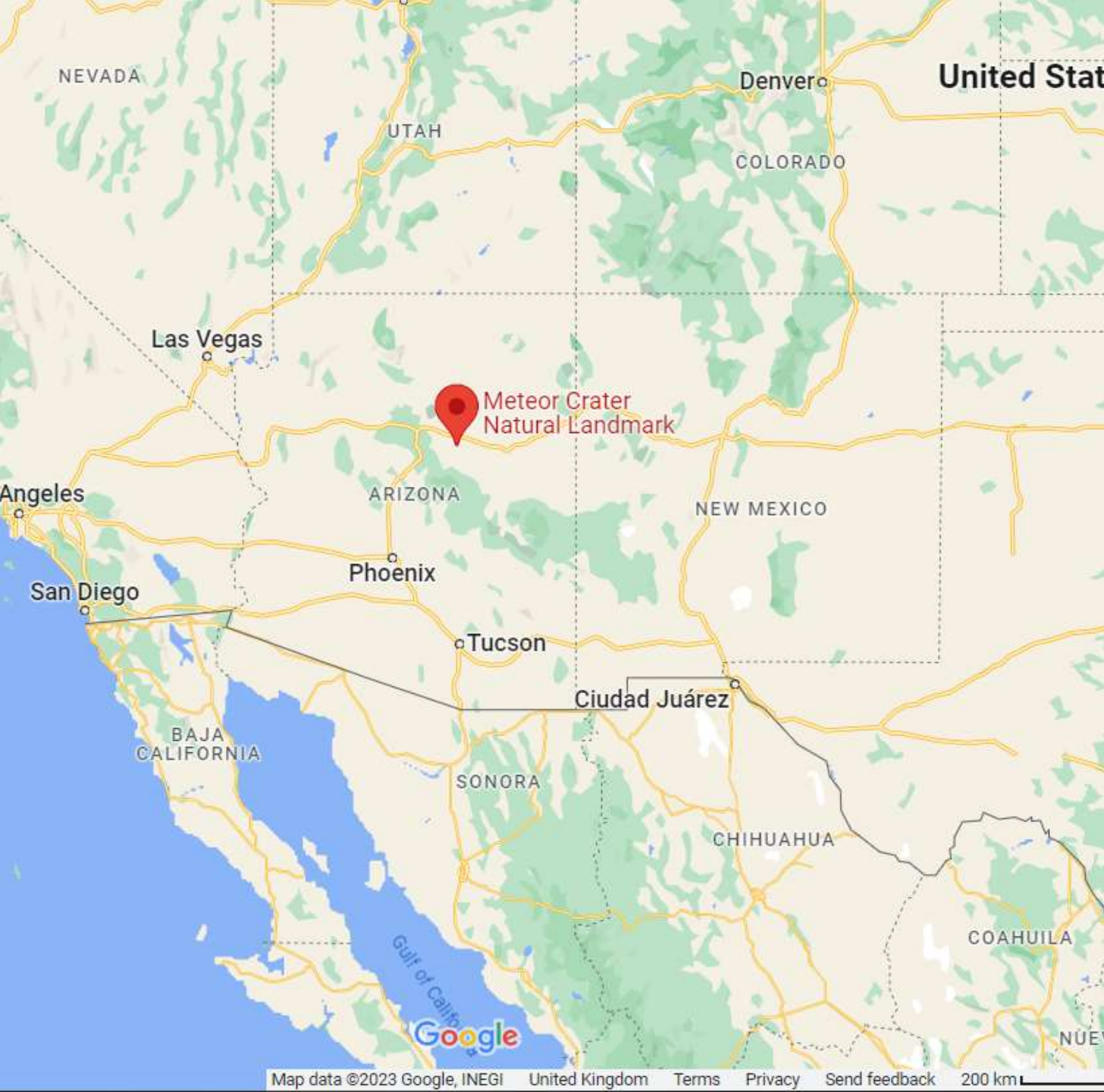




# The Barringer Meteor Crater Arizona

By

Mellissa Freeman

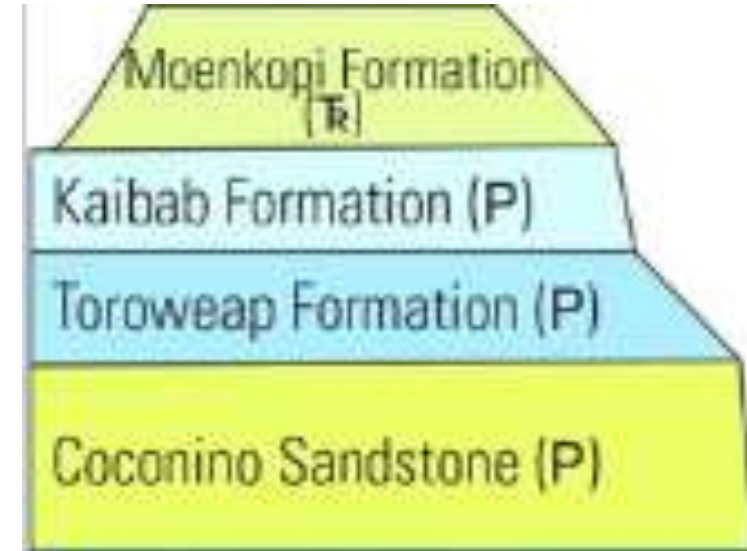
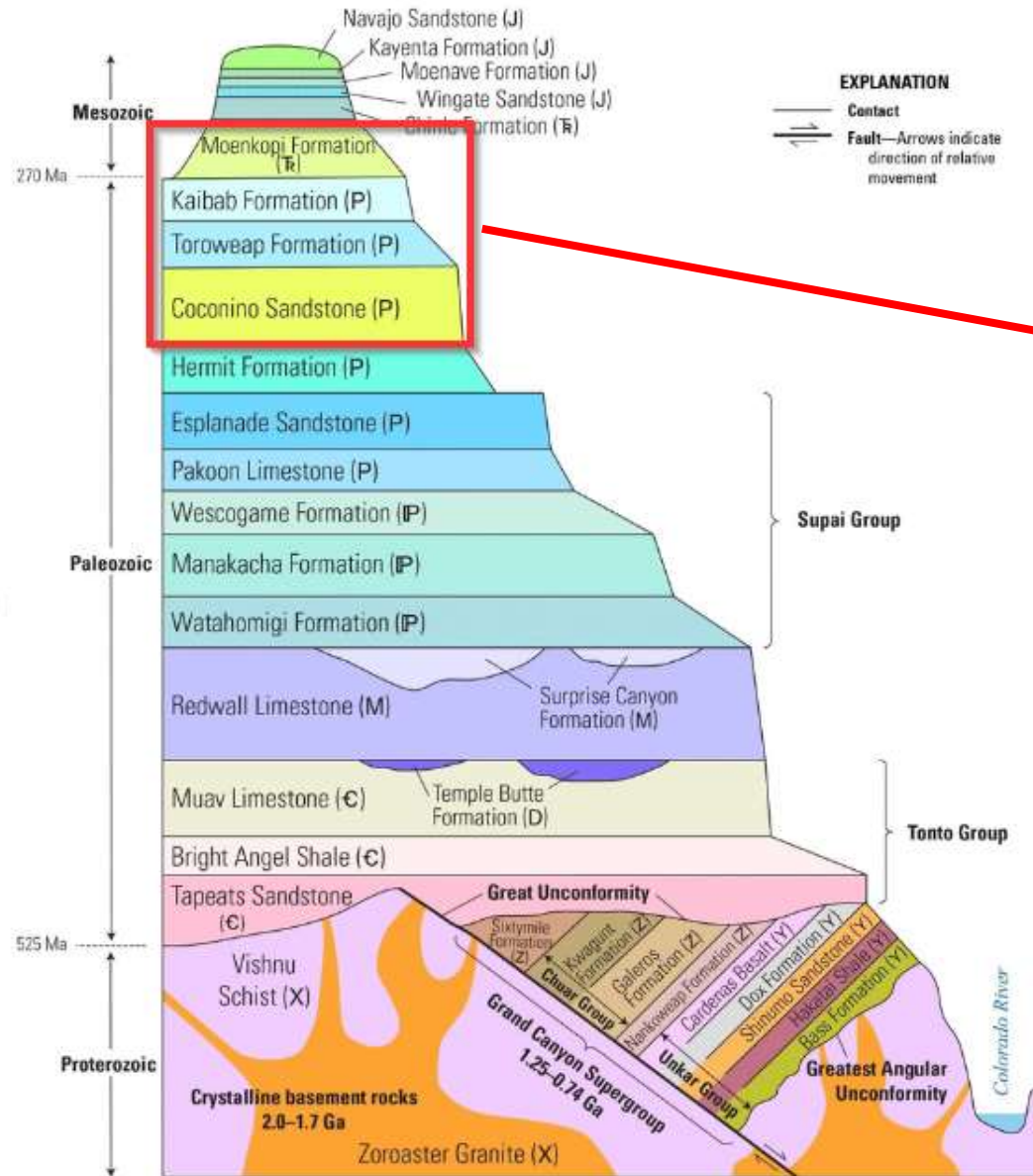


Location

Colorado Plateau



# The geology





# The history

- Impact happened approx. 50,000 years ago
- 1871 - The first written report was by a man by the name of Franklin
- 1886 – iron-nickel meteorites were found by a sheep herder
- 1891 – site visited by USGS who deemed it to be volcanic
- 1903 – site was purchased by Daniel Moreau Barringer
- 1941 – the Barringer family entered into a lease with Bar T Bar Ranch Company and formed the Meteor Crater Enterprises Inc which is still running today.















THE END OF PART 1....

# A brief look at naturally occurring fission reactors



# Uranium

- Actinide
  - Atomic number 92

V · T · E			Periodic table																
Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	Hydrogen & alkali metals	Alkaline earth metals											Triels	Tetrels	Pnictogens	Chalcogens	Halogens	Noble gases	
Period	Hydrogen 1 H 1.0080																	Helium 2 He 4.0026	
1																			
2	Lithium 3 Li 6.94	Beryllium 4 Be 9.0122											Boron 5 B 10.81	Carbon 6 C 12.011	Nitrogen 7 N 14.007	Oxygen 8 O 15.999	Fluorine 9 F 18.998	Neon 10 Ne 20.180	
3	Sodium 11 Na 22.990	Magnesium 12 Mg 24.305											Aluminum 13 Al 26.982	Silicon 14 Si 28.085	Phosphorus 15 P 30.974	Sulfur 16 S 32.06	Chlorine 17 Cl 35.45	Argon 18 Ar 39.95	
4	Potassium 19 K 39.098	Calcium 20 Ca 40.078	Scandium 21 Sc 44.956	Titanium 22 Ti 47.867	Vanadium 23 V 50.942	Chromium 24 Cr 51.996	Manganese 25 Mn 54.938	Iron 26 Fe 55.845	Cobalt 27 Co 58.933	Nickel 28 Ni 58.693	Copper 29 Cu 63.546	Zinc 30 Zn 65.38	Gallium 31 Ga 69.723	Germanium 32 Ge 72.630	Arsenic 33 As 74.922	Selenium 34 Se 78.971	Bromine 35 Br 79.904	Krypton 36 Kr 83.798	
5	Rubidium 37 Rb 85.468	Strontium 38 Sr 87.62	Yttrium 39 Y 88.906	Zirconium 40 Zr 91.224	Niobium 41 Nb 92.906	Molybdenum 42 Mo 95.95	Technetium 43 Tc [97]	Ruthenium 44 Ru 101.07	Rhodium 45 Rh 102.91	Palladium 46 Pd 106.42	Silver 47 Ag 107.87	Cadmium 48 Cd 112.41	Indium 49 In 114.82	Tin 50 Sn 118.71	Antimony 51 Sb 121.76	Tellurium 52 Te 127.60	Iodine 53 I 126.90	Xenon 54 Xe 131.29	
6	Caesium 55 Cs 132.91	Barium 56 Ba 137.33	*	Lutetium 71 Lu 174.97	Hafnium 72 Hf 178.49	Tantalum 73 Ta 180.95	Tungsten 74 W 183.84	Rhenium 75 Re 186.21	Osmium 76 Os 190.23	Iridium 77 Ir 192.22	Platinum 78 Pt 195.08	Gold 79 Au 196.97	Mercury 80 Hg 200.59	Thallium 81 Tl 204.38	Lead 82 Pb 207.2	Bismuth 83 Bi 208.98	Polonium 84 Po [209]	Astatine 85 At [210]	Radon 86 Rn [222]
7	Francium 87 Fr [223]	Radium 88 Ra [226]	**	Lawrencium 103 Lr [260]	Rutherfordium 104 Rf [267]	Dubnium 105 Db [268]	Seaborgium 106 Sg [269]	Bohrium 107 Bh [270]	Hassium 108 Hs [280]	Meitnerium 109 Mt [288]	Darmstadtium 110 Ds [281]	Roentgenium 111 Rg [285]	Copernicium 112 Cn [285]	Nihonium 113 Nh [286]	Flerovium 114 Fl [289]	Moscovium 115 Mc [290]	Livermorium 116 Lv [293]	Tennesseine 117 Ts [294]	Oganesson 118 Og [294]
			*	Lanthanum 57 La 138.91	Cerium 58 Ce 140.12	Praseodymium 59 Pr 140.91	Neodymium 60 Nd 144.24	Promethium 61 Pm [145]	Samarium 62 Sm 150.36	Europium 63 Eu 151.96	Gadolinium 64 Gd 157.25	Terbium 65 Tb 158.93	Dysprosium 66 Dy 162.50	Holmium 67 Ho 164.93	Erbium 68 Er 167.26	Thulium 69 Tm 168.93	Ytterbium 70 Yb 173.05		
			**	Actinium 89 Ac [227]	Thorium 90 Th 232.04	Protactinium 91 Pa 231.04	Uranium 92 U 238.03	Neptunium 93 Np [237]	Plutonium 94 Pu [244]	Americium 95 Am [243]	Curium 96 Cm [247]	Berkelium 97 Bk [247]	Californium 98 Cf [251]	Einsteinium 99 Es [252]	Fermium 100 Fm [257]	Mendelevium 101 Md [258]	Nobelium 102 No [259]		

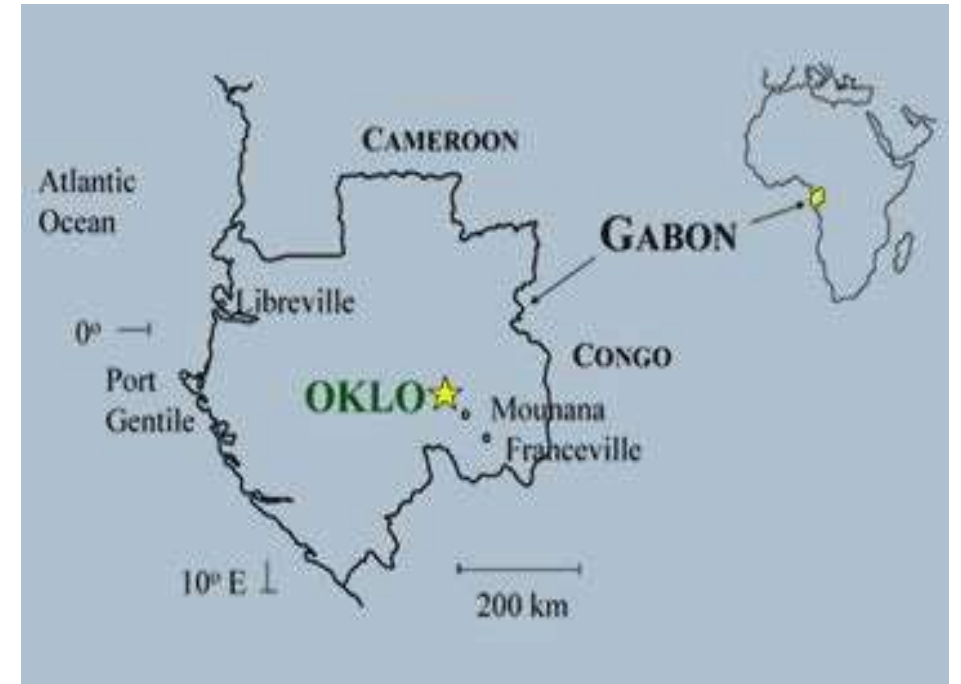
# Uranium

- 2 isotopes
  - $^{235}\text{U}$
  - $^{238}\text{U}$
- Both radioactive but half lives differ
  - $^{235}\text{U}$ : 700 million years
  - $^{238}\text{U}$ : 4.6 billion years
- Natural relative abundance
  - $^{235}\text{U}$ : 0.72%
  - $^{238}\text{U}$ : 99.28%
  - Relative abundance is remarkably consistent worldwide...



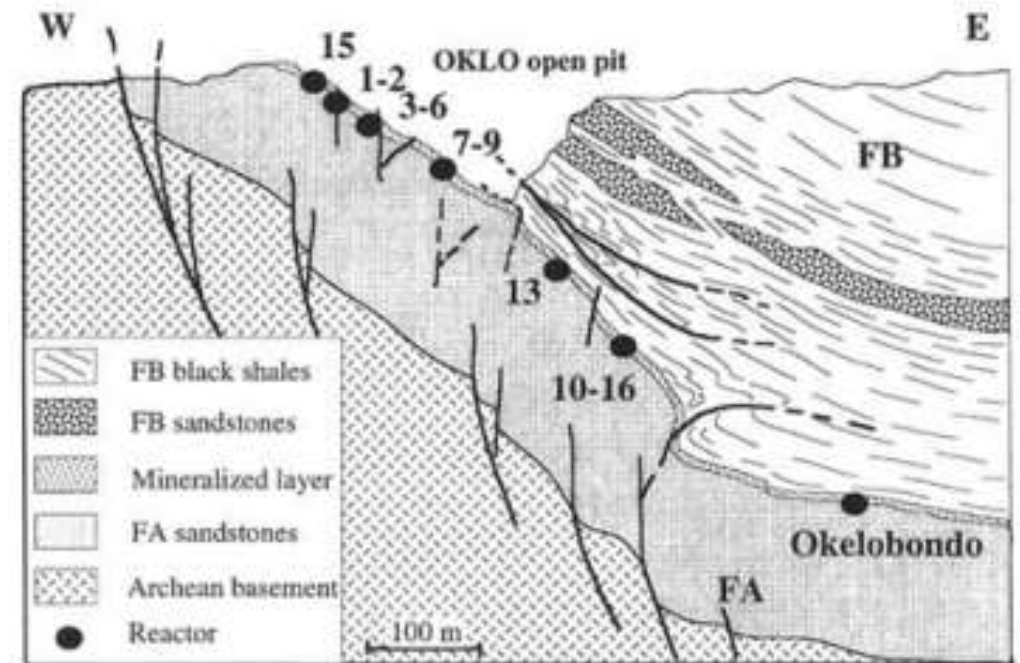
# Relative abundance

- ...except in zones in the uranium mines at Oklo, Okelobondo and Bangombé in the Franceville Basin, Gabon
- In these zones
  - $^{235}\text{U}$  is low as 0.38 %
  - an assembly of the end-member fission products which result from sustained nuclear reactions in a nuclear reactor



# Geological setting

- Uraninite deposits
  - located near the top of a succession of sandstones and conglomerates
- Uraninite deposits
  - Formed by dissolution-precipitation processes
  - From which approximately 28,000 tonnes of uranium ore have been mined
- Age:  $1950 \pm 40$  Ma - Proterozoic



*Gauthier-Lafaye et al 1996*



# Geological setting

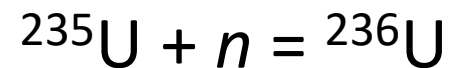
- U-bearing conglomerates
  - Buried by deltaic marine deposits of Upper FA formation and FB shales
  - U(VI) oxides
- Shales were rich in organic matter
  - Burial at depths of up to 4 km
  - Temperature and pressure conditions of “oil window” causing conversion to hydrocarbons
  - Hydrocarbons
    - migrated to the FA sandstone reservoir rock
    - Accumulated in structural traps

# Geological setting

- Uranium mineralisation
  - fluids containing U(VI) oxides meet fluids bearing hydrocarbons
  - hydrocarbons reduce U(VI) oxides to U(IV) oxides
- Uranium is precipitated as uraninite ( $\text{UO}_2$ ) in the sandstone in
  - pores
  - fractures
- Zones
  - uraninite seams are depleted in uranium but contain fission products

# Nuclear fission

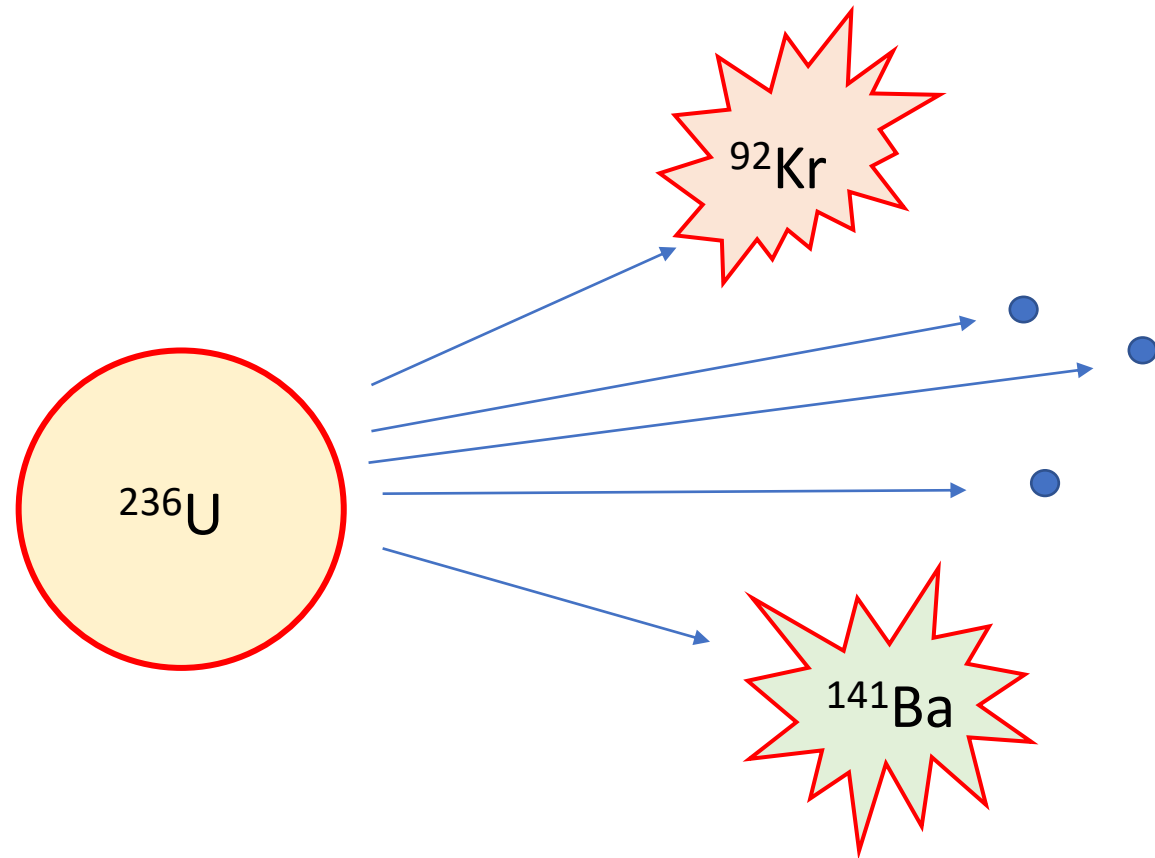
- Inherently unstable nuclei of radioactive isotopes “decay” spontaneously emitting a neutron
- Decay occurs more readily if provoked by disturbance of the structure of the nucleus
- Most readily caused by the absorption of a neutron
- Certain isotopes, termed “fissile”, will absorb a neutron
  - $^{235}\text{U}$  is fissile
  - $^{238}\text{U}$  is not fissile





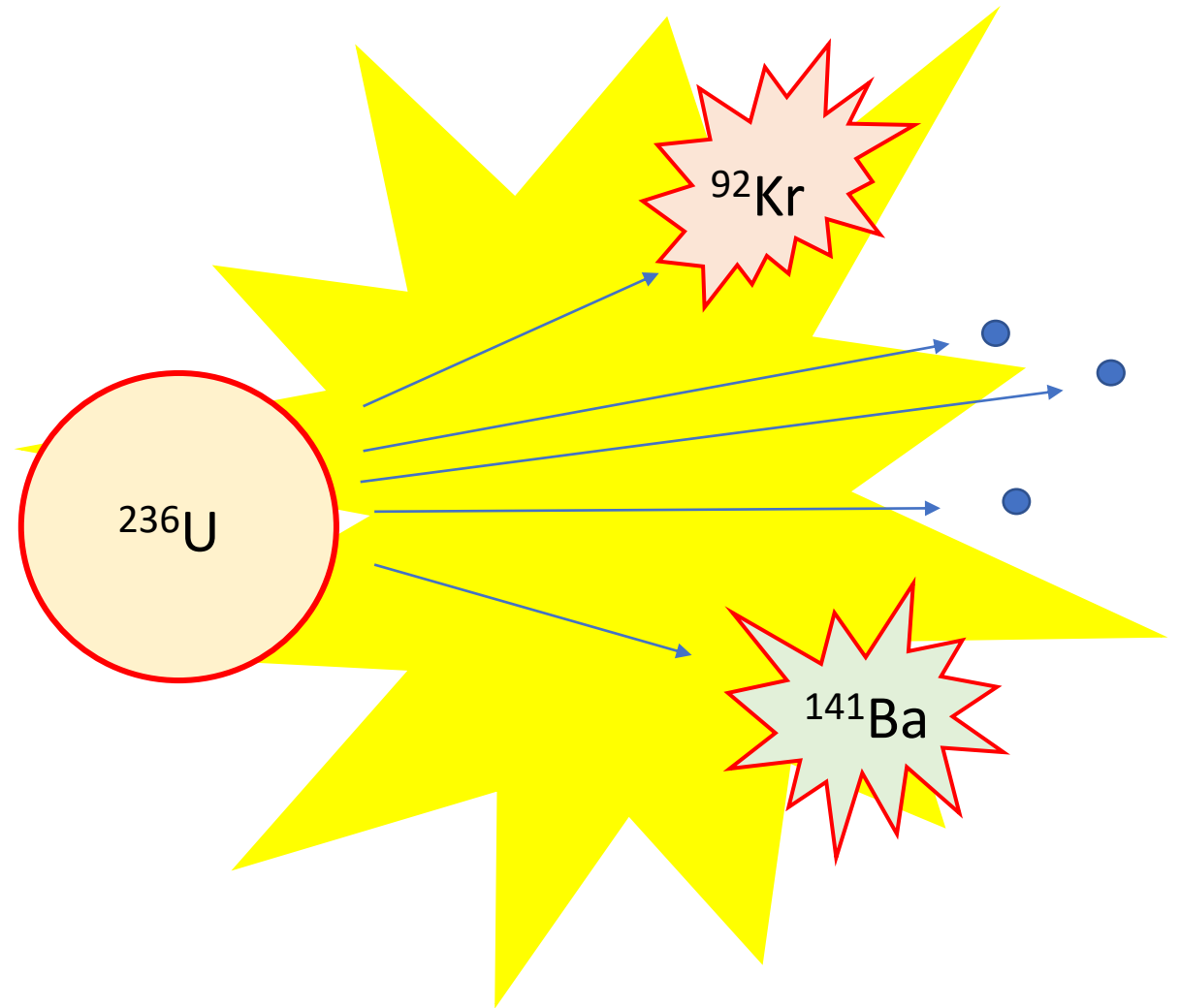
# Nuclear fission

- The fissile nucleus becomes even more unstable and splits into
  - 2 fission nuclei
  - 2 to 3 free neutrons travelling at high speed



# Nuclear fission

- Total mass of parts is very slightly less than original atom
- Missing mass has been converted into energy (released in the form of heat)
  - $E = mc^2$



# Nuclear fission

- These neutrons may go on to trigger fission of neighbouring fissile isotopes – a chain reaction
- A chain reaction may be
  - Uncontrolled – runaway nuclear reaction – basis of atomic weapons or
  - Controlled (“moderated”) - basis of nuclear power
- For a controlled nuclear reaction, neutron speed is crucial
  - Too fast – will go through nuclei = no neutron capture – chain reaction shuts down
  - Too slow – insufficient energy = no neutron capture – chain reaction shuts down
  - Goldilocks speed = neutron capture followed by fission



# Nuclear fission

- High speed neutrons
  - inelastic collisions with things of similar mass will slow down neutrons emitted by fission
  - In liquid form, some water molecules dissociate into hydrogen ions and oxygen ions
  - A hydrogen ion has the same mass as a neutron
  - Water is a good moderator

# Criteria for self-sustaining chain fission reactions i.e. a nuclear reactor

1. Sufficient fuel
2. Sufficiently high ratio of  $^{235}\text{U}$ :  $^{238}\text{U}$
3. Containment
4. Moderator

# Conditions at Oklo

## 1. Sufficient fuel

- U ore seams are at least 0.5 m and typically contained in layers 2m thick
- Sufficient concentration of uranium-bearing minerals
  - Minimum requirement of 10%
  - In reactor zones, typically 20 to 87%



Remains of reactor at location 2



# Conditions at Oklo

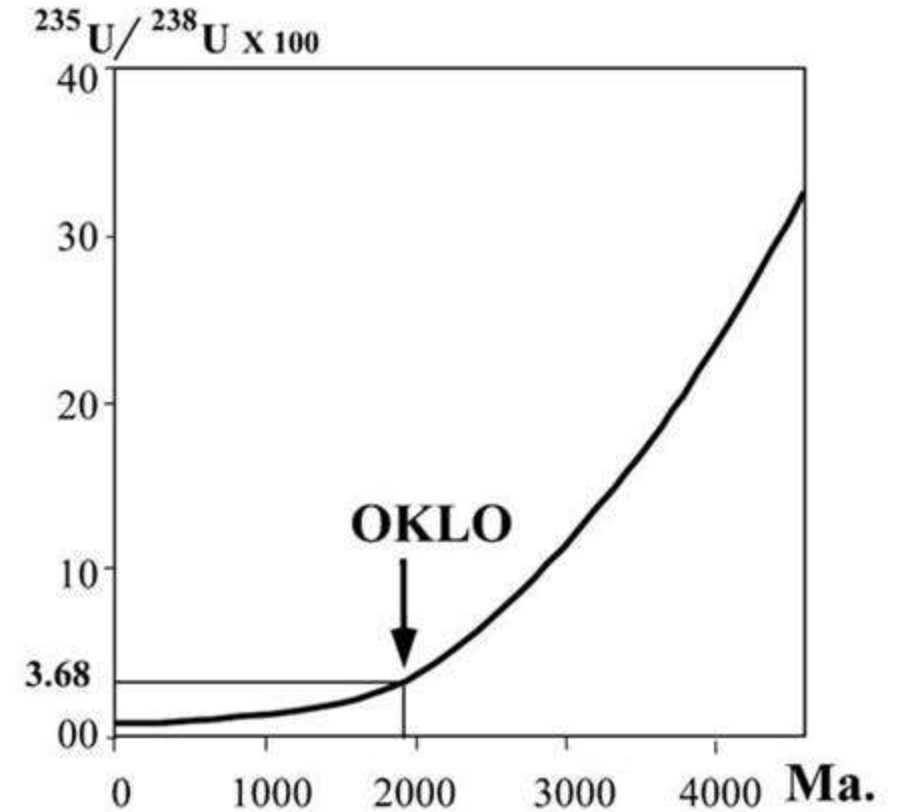
## 2. Sufficiently high ratio of $^{235}\text{U}$ : $^{238}\text{U}$

- Based on the half lives, extrapolation backwards to nearly 2Ga indicates that the relative abundance was approximately

U – 235: 3.68%

U – 238: 96.32%

- Fuel for nuclear power plant: approximately U – 235: 3 to 5%

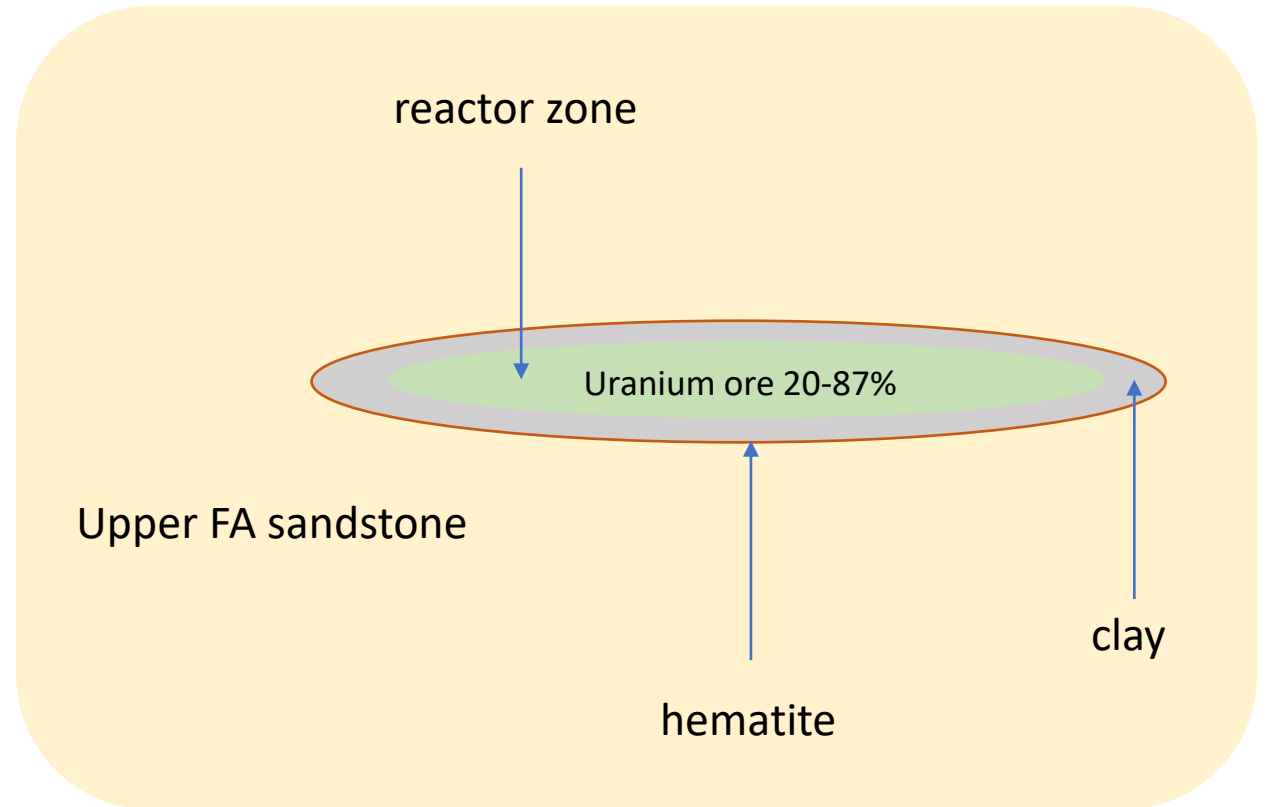


*Gauthier-Lafaye 2002*

# Conditions at Oklo

## 3. Containment

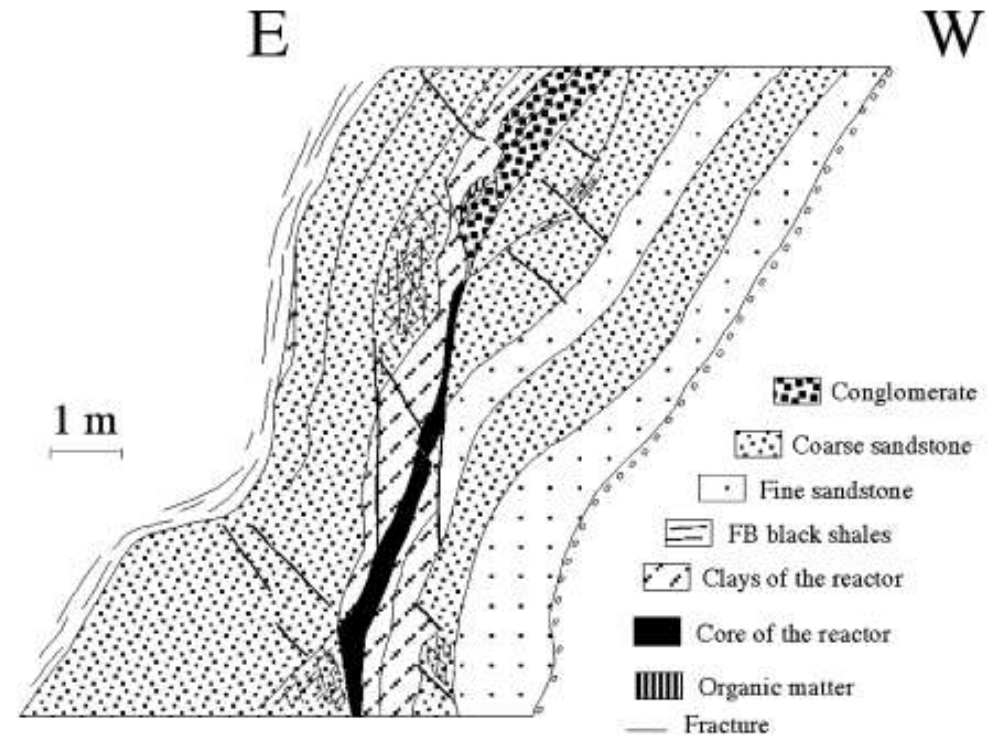
- Enveloped by clay minerals
  - e.g. illite
- Quartz-rich sandstone are natural neutron reflectors
- Sharp boundary between reactor zone and underlying sandstone
  - often with a thin layer of hematite ( $\text{Fe}_2\text{O}_3$ )



# Conditions at Oklo

## 4. Moderator

- Meteoric water seeping down through porous rocks and faults



Rector 9 from earlier figure *Gauthier-Lafaye et al 1996*



# Duration of Oklo fission reactors

- Short-term
  - Heat generated by the fission reactions caused the water to boil away
  - Without a moderator, the chain reactions shut down
  - When cooled sufficiently for water to flow back, neutron speed is moderated again and chain reactions recommence
  - Calculations indicate
    - Activity – approximately 30 minutes
    - Shut down – approximately 2 hours 30 minutes
    - i.e. a 3 hour cycle
- Long-term
  - Differs between reactor sites – approximate range 62,000 to 270,000 years
  - Analyses of abundances of fission products

# Depleted Uranium

- Research into current conditions at Oklo - useful analogue in connection with disposal of nuclear fuels

# References

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- Janeczek, J. *Mineralogy and Geochemistry of Natural Fission Reactors in Gabon. Reviews in Mineralogy Volume 38 p.321*
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- Gauthier-Lafaye, F. (2002). "2 billion year old natural analogs for nuclear waste disposal: the natural nuclear fission reactors in Gabon (Africa)". *Comptes Rendus Physique*. **3** (7–8): 839–849. *Bibcode:2002CRPhy...3..839G*. *doi:10.1016/S1631-0705(02)01351-8*.

THE END OF PART 2....



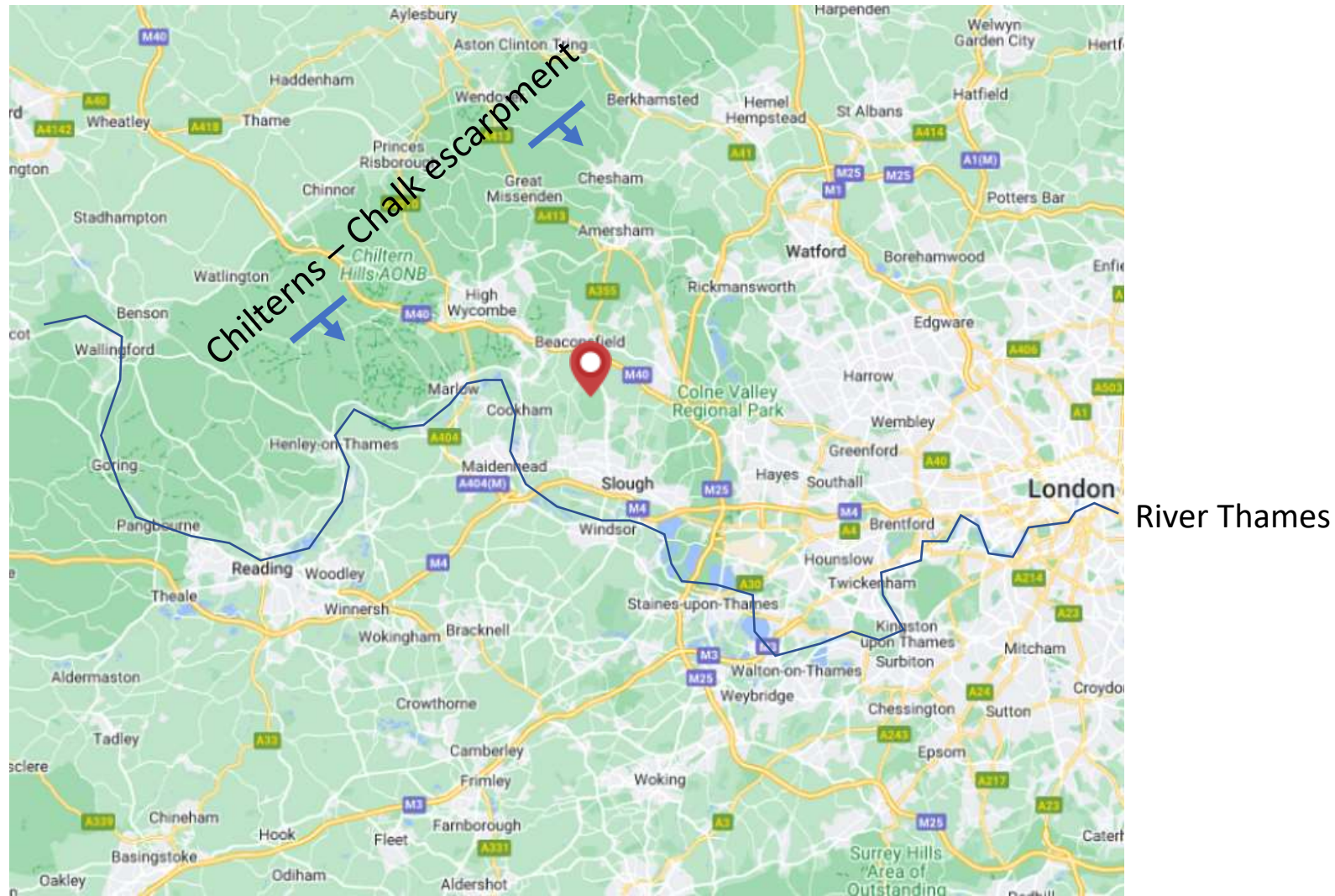
# Burnham Beeches Hydrology

Bath Geological Society

Graham Hickman

May 2023

# Burnham Beeches – 25miles west of London



Northern Edge of Thames Valley, Western Edge of London Basin





Ancient Beech Pollards – SSSI, thin sandy soils

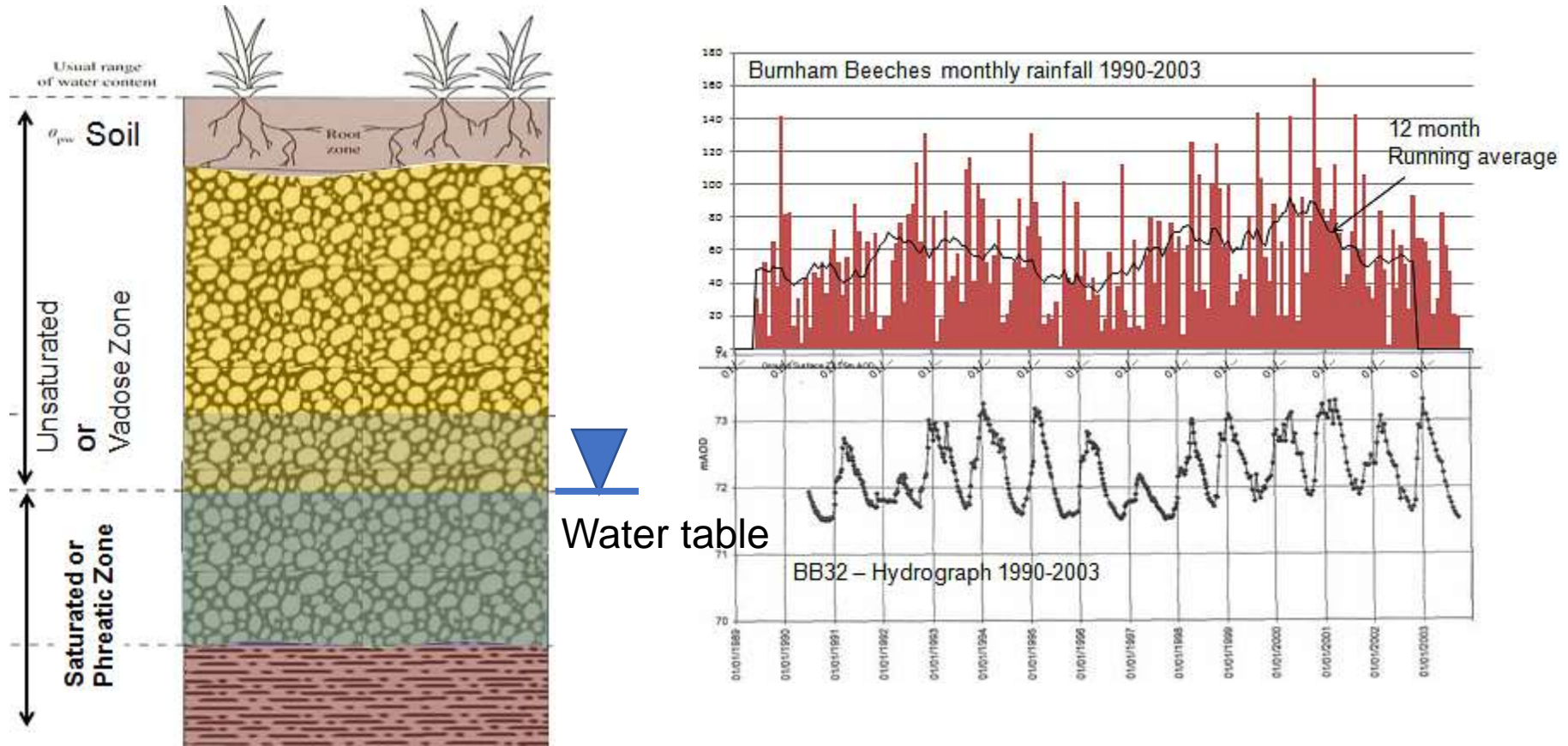




Water observation bore holes



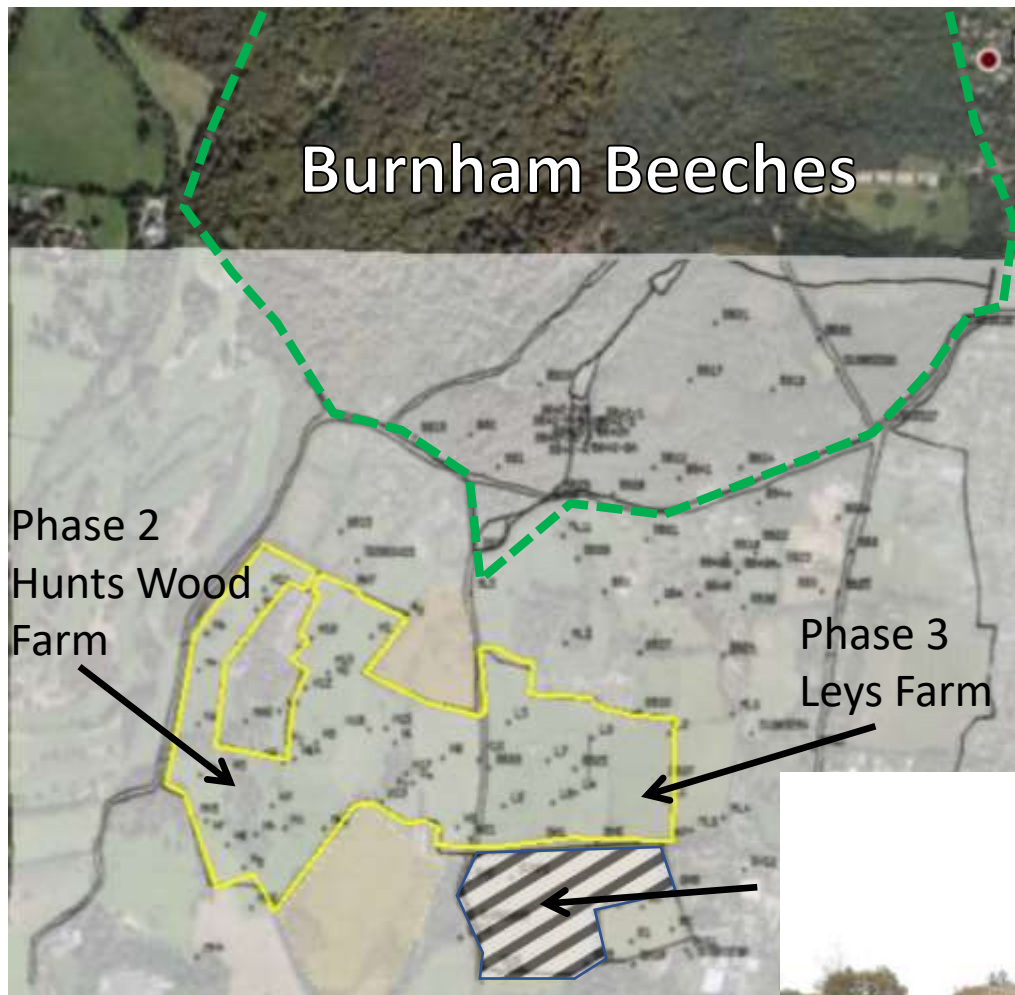
# Hydrology the Basics



During summer – less rain, more evaporation, take-up by plants – water levels fall

During winter – more rain, less evaporation, reduced plant take-up – water levels rise  
more run-off, bogy areas and springs more active





## East Burnham Quarry

Despite strong opposition, in 1982 permission was given to extract gravels from a lower Terrace (Boyn Hill) south of Burnham Beeches.

Concerns about impact on Burnham Beeches by lowering water table. If a hydraulic connection exists. In 1988 this resulted in a program of boreholes being drilled and monitored.



Based on borehole BB44. ( SU 95367 84326) near to Stag Public House



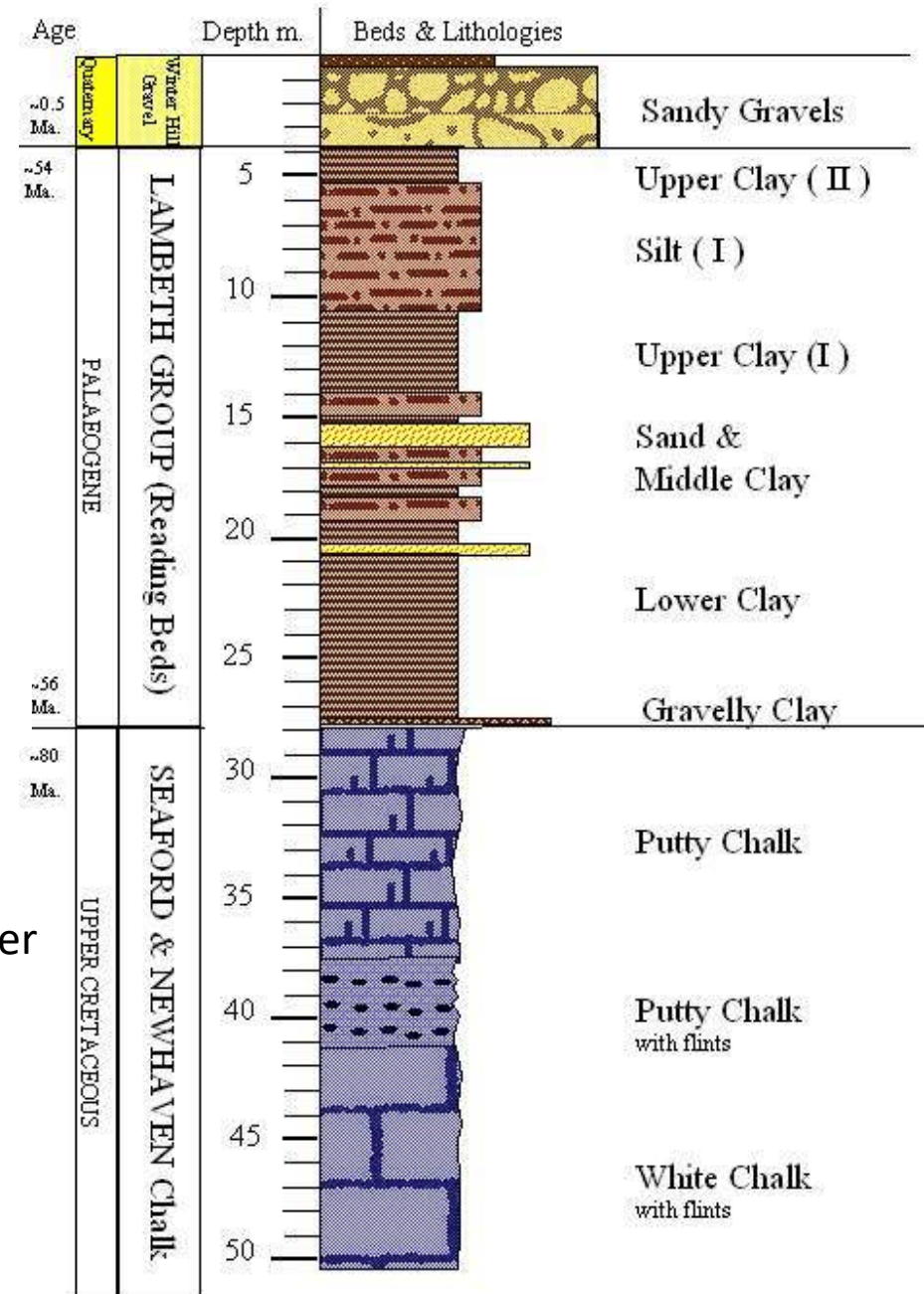
Winterhill  
Terrace  
Quaternary  
~0.45Ma



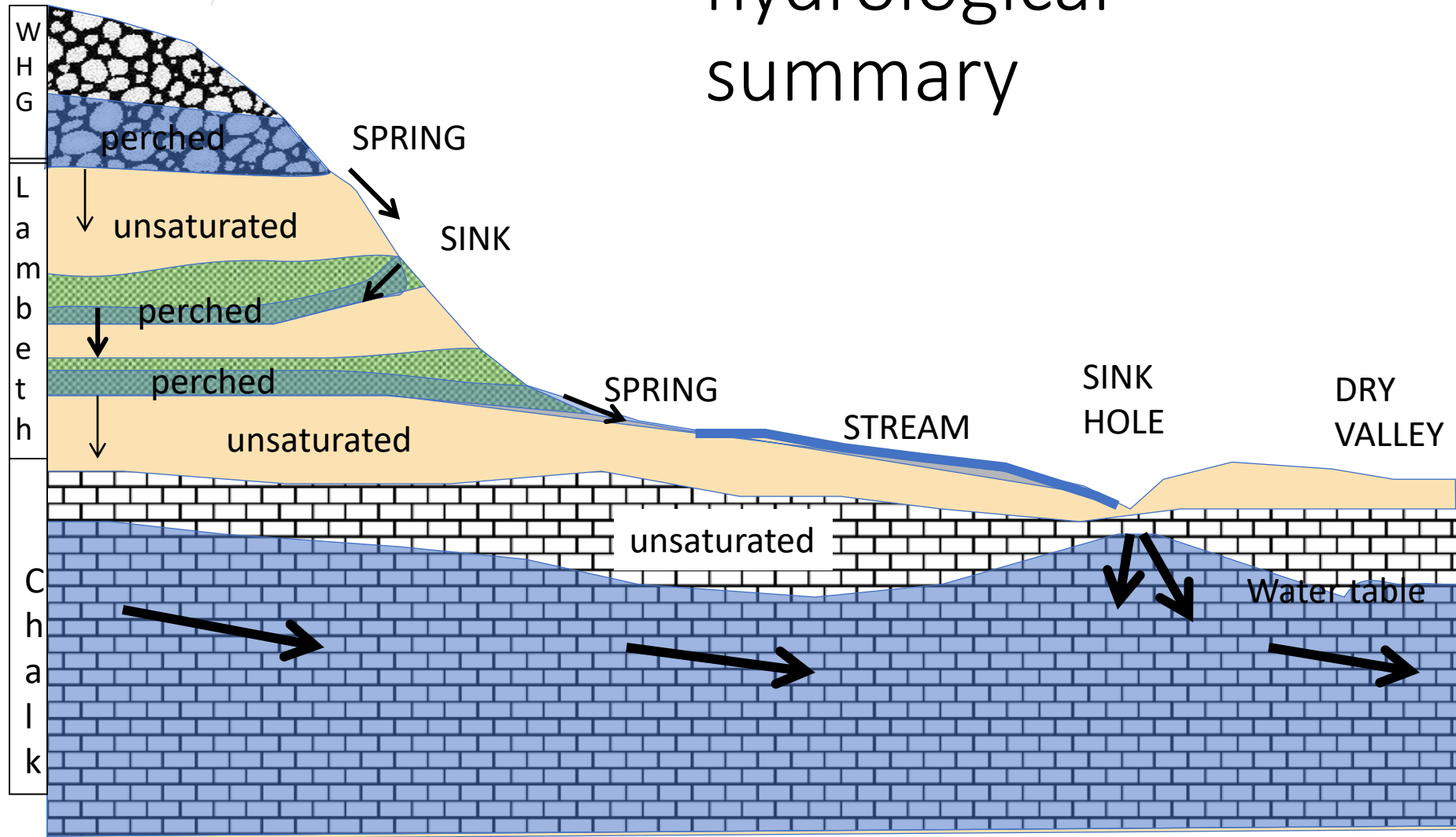
Lambeth  
Group  
Palaeocene  
56-54Ma



Chalk – Upper  
Cretaceous  
~80Ma

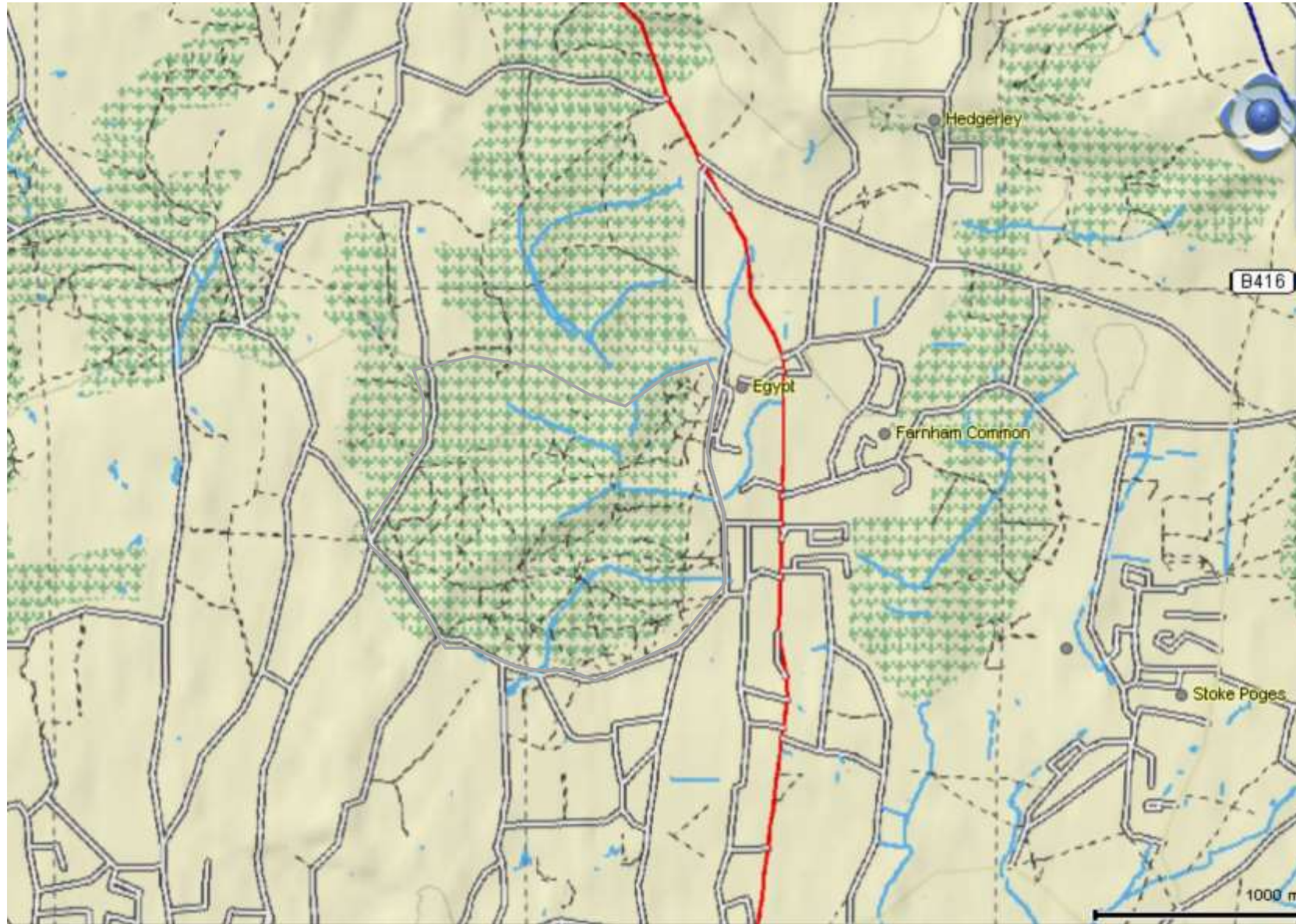


# Burnham Beeches hydrological summary





# Streams in and around Burnham Beeches







Springs develop at interface between gravels and clays, notice also break in slope





Clays and silts – springs, mires, streams.





Swilly Pond -May 9<sup>th</sup> 2010

Swilly Pond May 9<sup>th</sup> 2010





Swilly Pond -Jan 3<sup>rd</sup> 2010

Swilly pond Jan 3<sup>rd</sup> 2010

A photograph of a cluster of white mushrooms growing on a tree trunk. The mushrooms have smooth, slightly convex caps and gills. The tree bark is dark and textured. In the background, there are blurred green leaves and tree trunks, suggesting a forest setting. The lighting is soft, highlighting the mushrooms against the dark background.

Thank you !

I hope this talk has inspired you to visit

THE END OF PART 3.....



# A Bit of Brazil In Bath

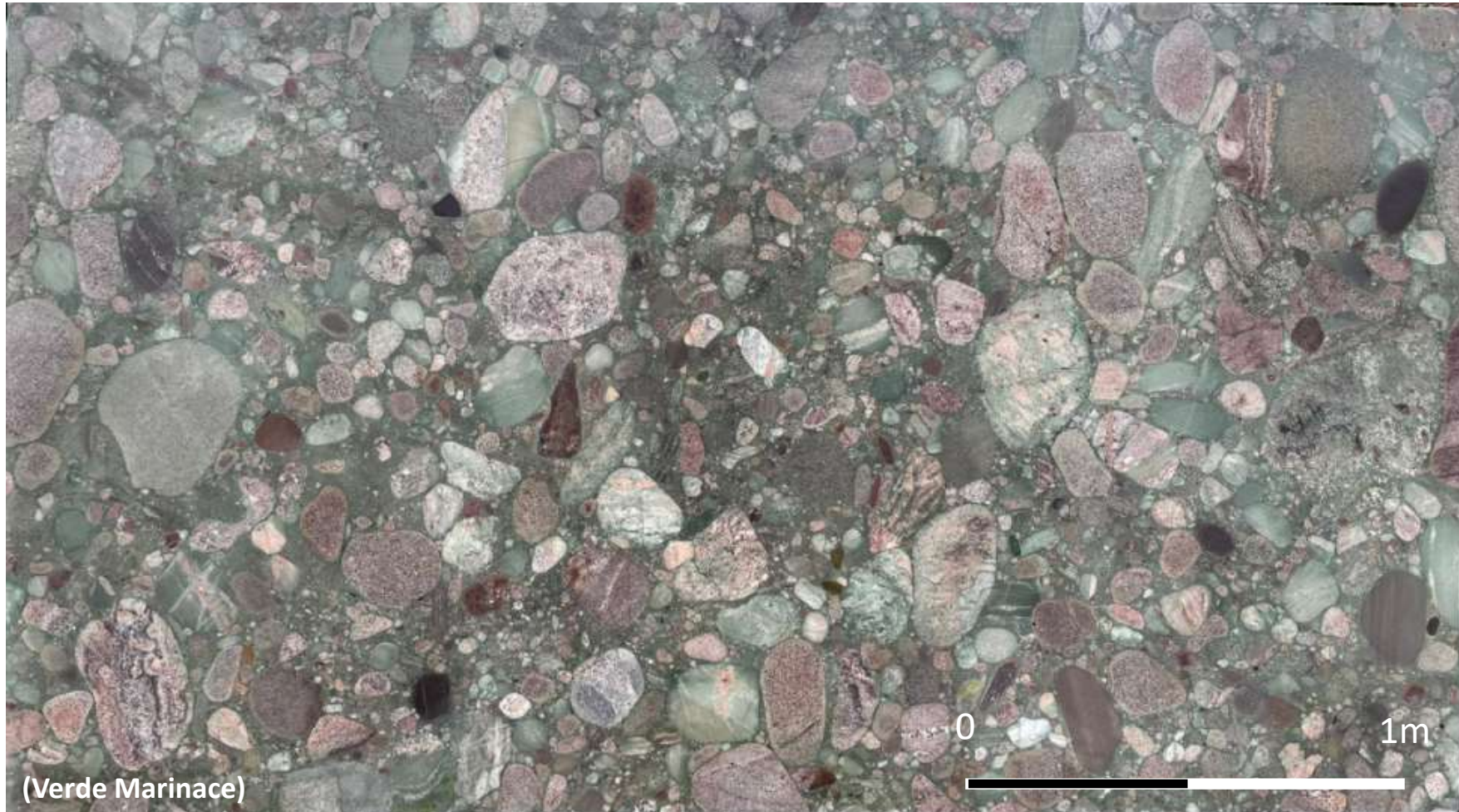
Bath Christmas Social 2021 (Geology of Kitchen Worktops).

★ Star of The Show....

**Paleoproterozoic polymict metaconglomerate** (Eastern Brazil)



On public display in Bath.....





## Varieties...



(Negro Marinace)



(Rosso Marinace)

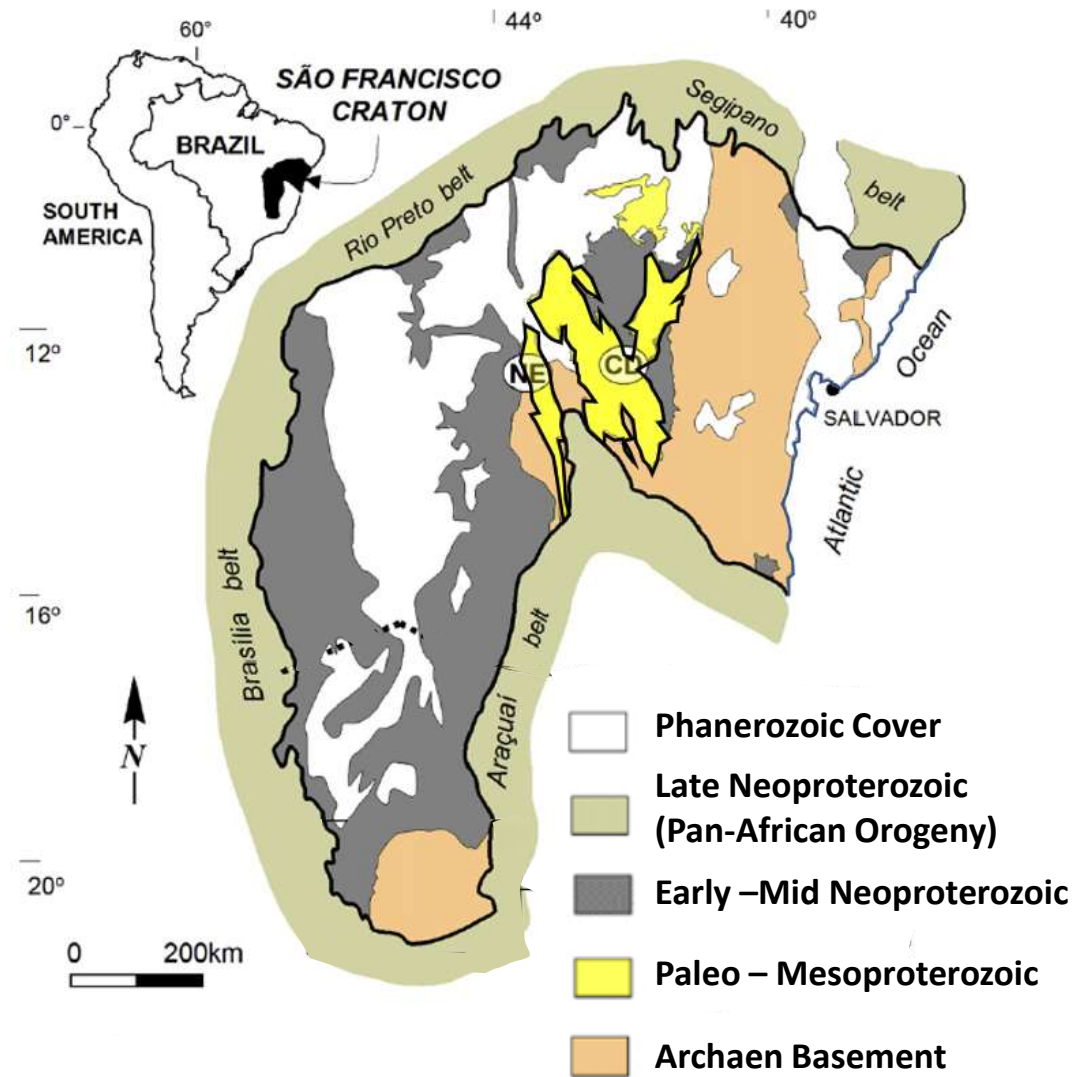
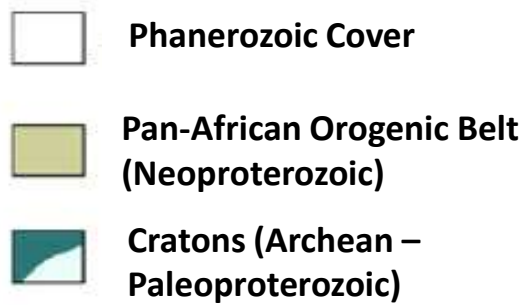


(Mixed)

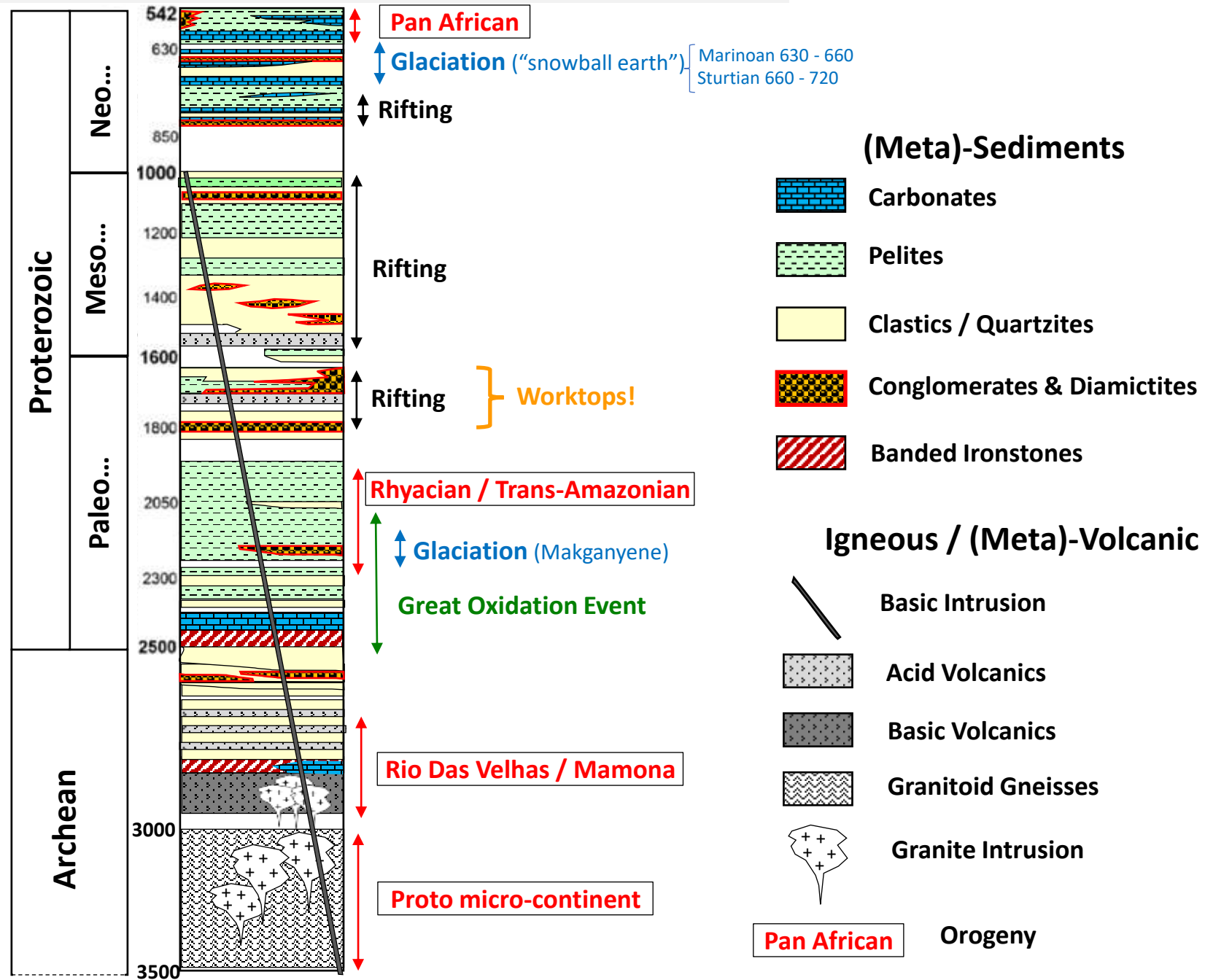
## University College Hospital London (Monolith and Shadow sculpture by John Aitken)



# Geological Context



# Sao Francisco Craton : Precambrian Sequence (simplified)





# Paleoproterozoic Rift Geometries (schematic)

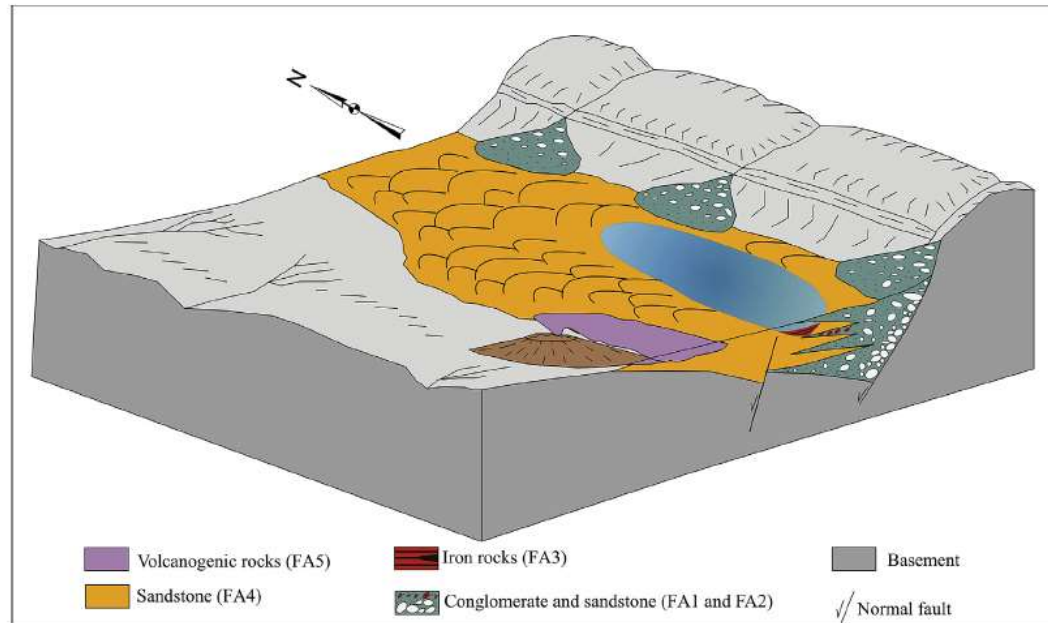
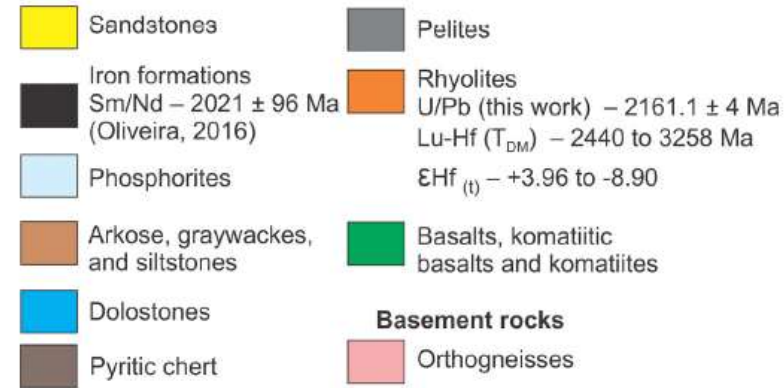
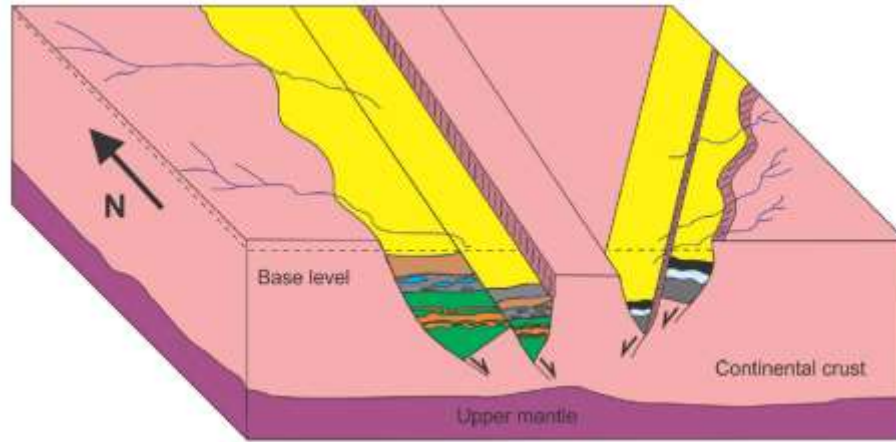
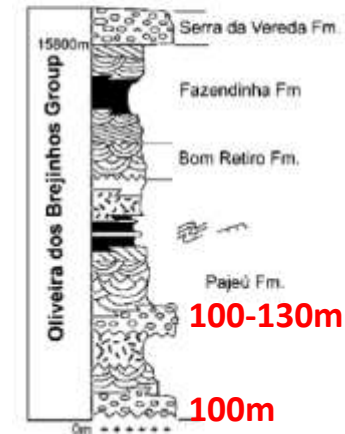
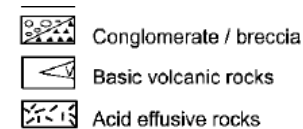
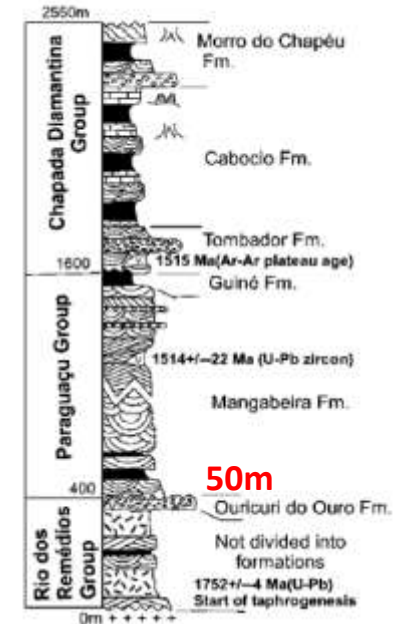


Fig. 16. Schematic illustration of the basin-fill evolution for the Terra Vermelha Group and related facies association.

## Northern Espinhaço range



## Chapada Diamantina



# Metasediment Outcrops

a) Actinolite Marble



b) Metadolostone (showing bedding)

c) Metachert (pyritic massive)



d) Phyllite

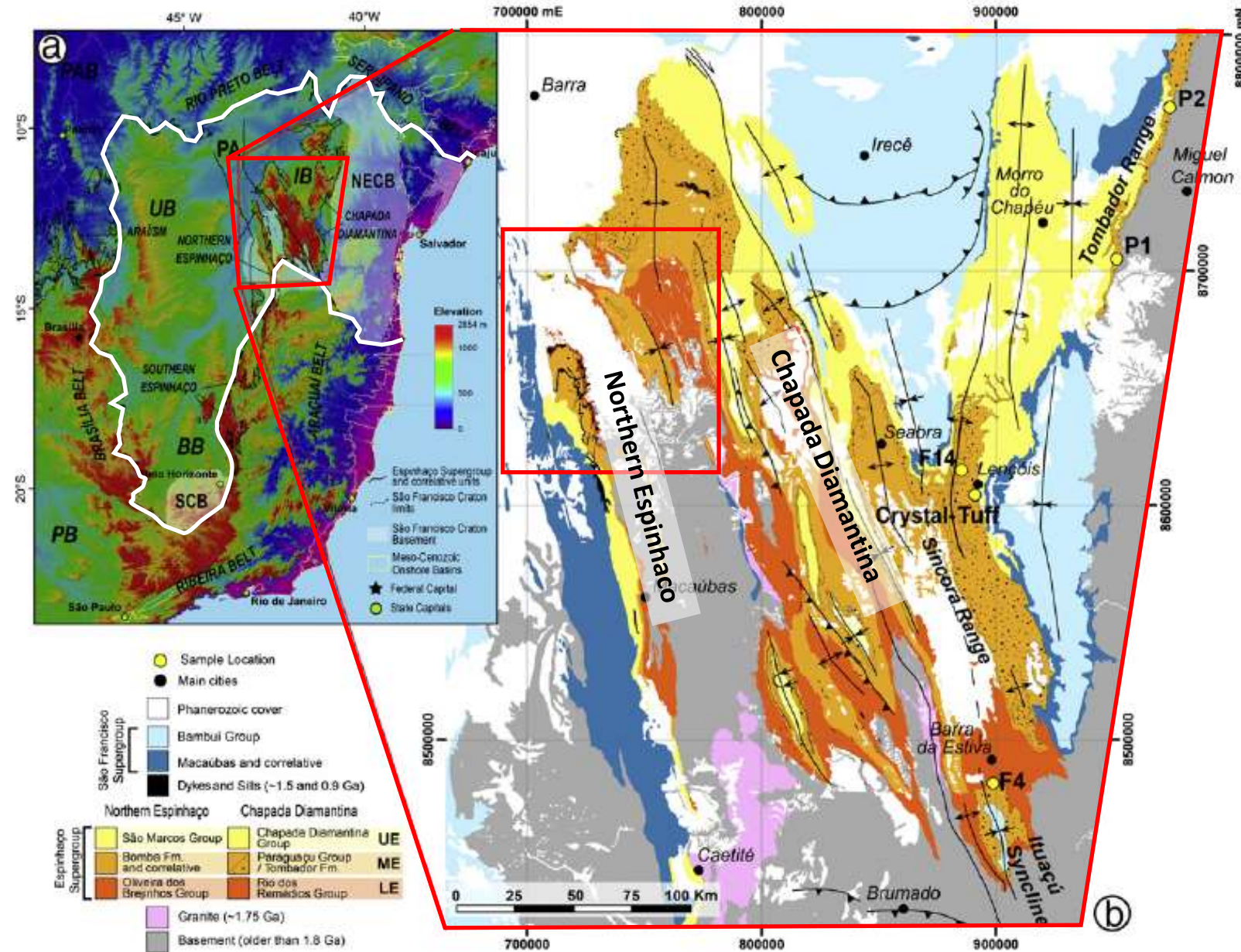
e) Quartzite with depositional with light (quartz-rich) and dark (tremolite & tourmaline-rich) layers



f) Magnetite-bearing banded iron formation

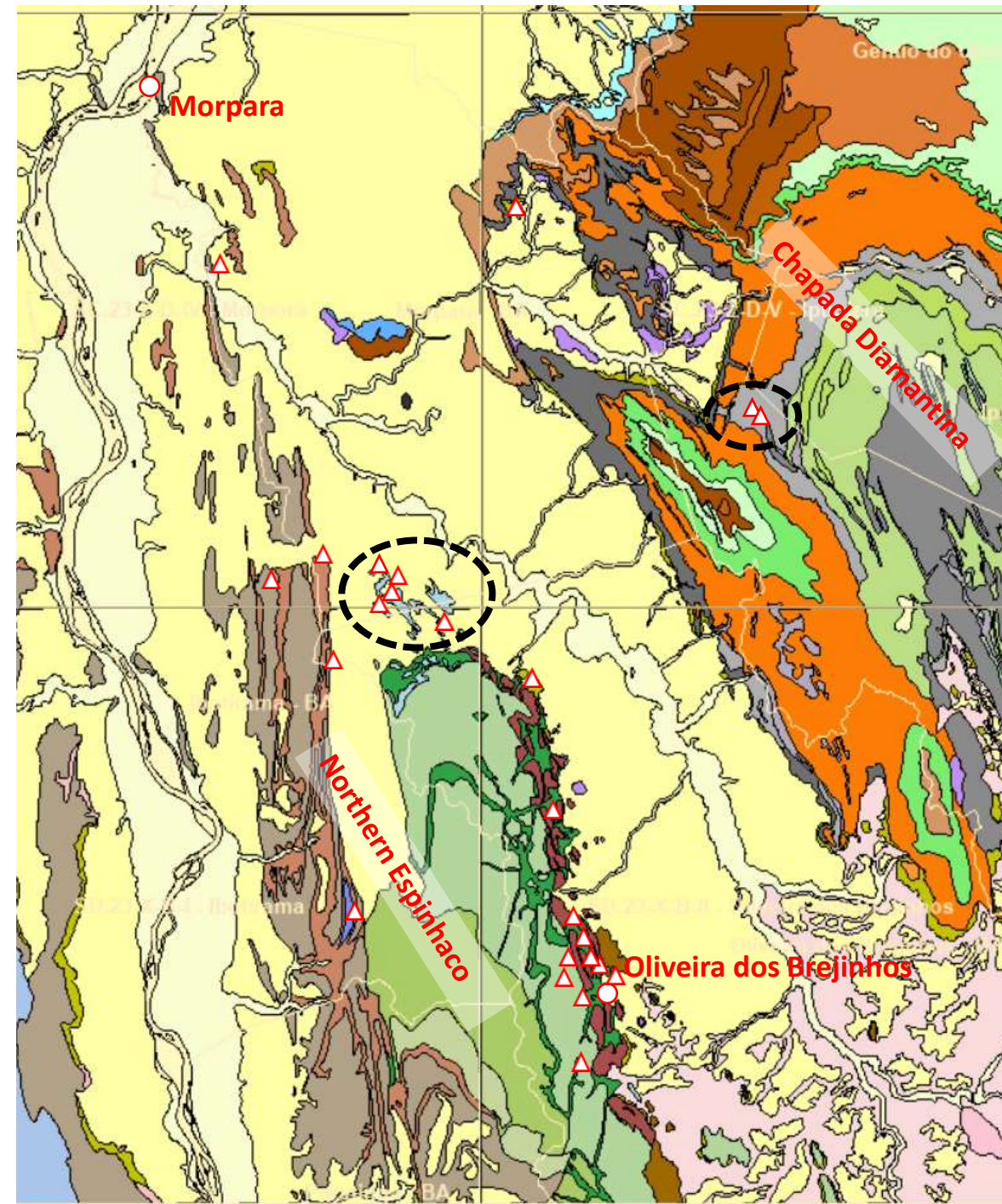
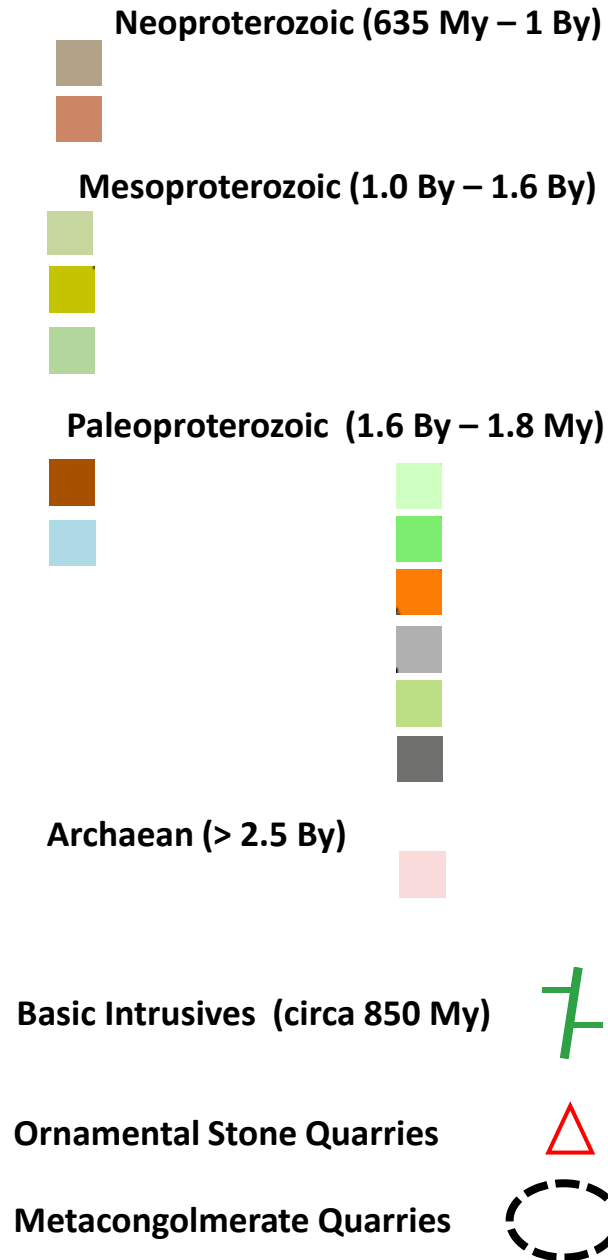


# Area of Interest (Northern Espinhaco & Chapada Diamantina)





# Surface Geology / Quarries





A bit of Brazil In Bath can be found at....



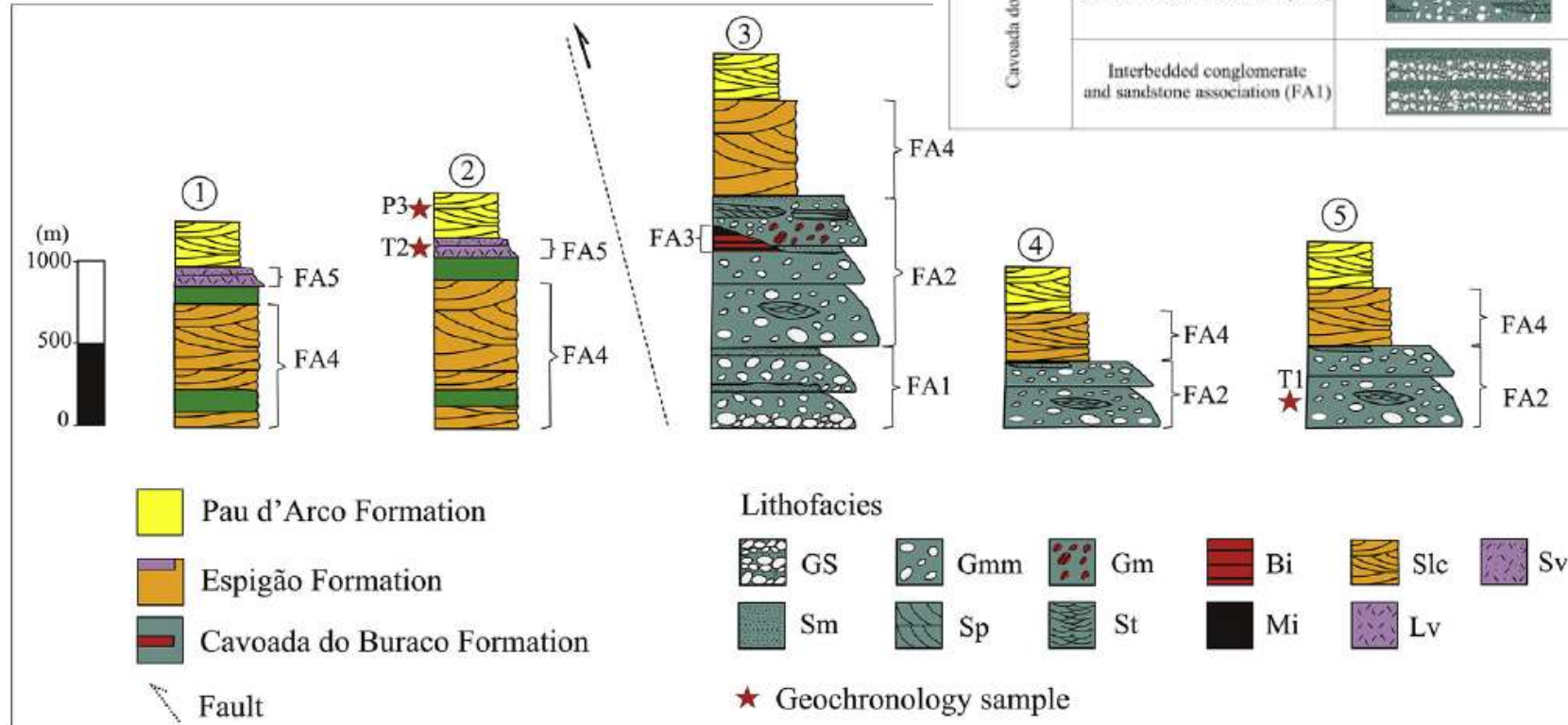




**Extras....**



# Record of a Statherian rift-sag basin in the Central Espinhaço Range: Facies characterization and geochronology. Alice Fernanda de Oliveira Costa (2018)



Lithostratigraphy	Facies association	Facies architecture and geometry	Lithofacies
Espigão Formation	Volcanogenic lithofacies (FA5)		Lv, Sv
	Large-scale cross-stratified sandstone succession (FA4)		Slc
Cavoadá do Buraco Formation	Iron Formation association (FA3)		Bi, Mi
	Clast to matrix-supported conglomerate with minor sandstone lens association (FA2)		Gmm, Gm, Sm, Sp, St, Slc
	Interbedded conglomerate and sandstone association (FA1)		GS

**Fig. 3.** General stratigraphic columns showing disposition of the different facies association and stratigraphic units.



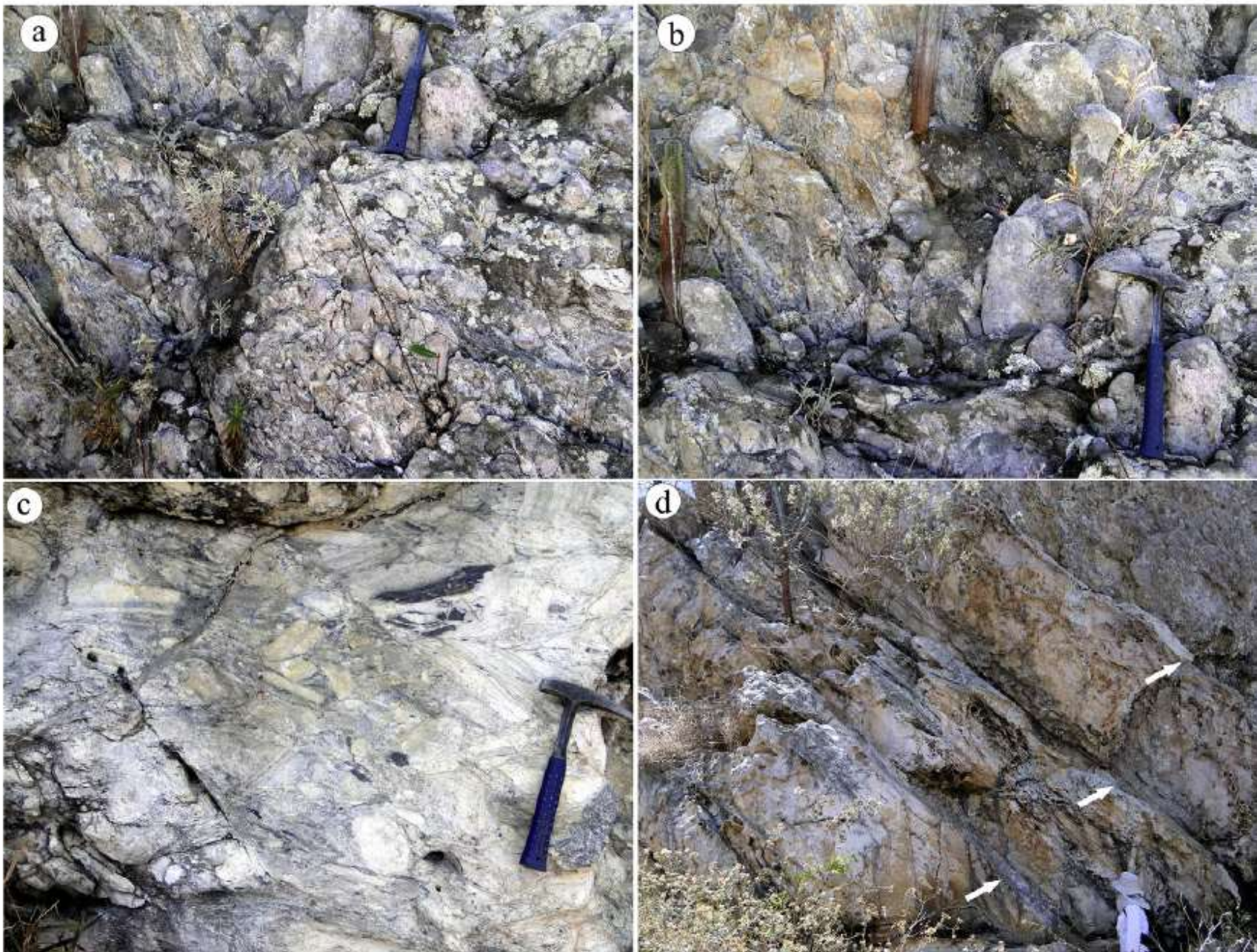


Fig. 5. Lithofacies from FA1 association: (a) Massive matrix-to clast-supported conglomerate with disorganized framework; (b) layer of matrix-supported with scattered boulders; (c) layer of clast-supported conglomerate with angular clasts; (d) matrix-supported conglomerate separated by thin layers of sandstone (Sm).





**Fig. 7.** Lithofacies from FA2 association: (a) Conglomerate Gmm with a sandstone St lens; (b) Matrix-supported polymictic conglomerate (Gm lithofacies); (c) Matrix-supported conglomerate with Bif's clasts; (d) Large-scale cross-stratified sandstone (Slc lithofacies) with planar-cross bedded sandstone bed (lithofacies Sp); (e) Enlarged view of lithofacies Sp marked by heavy minerals; (f) Enlarged view of lithofacies Slc.



Sequence stratigraphy of the mixed wave-tidal-dominated Mesoproterozoic sedimentary succession in Chapada Diamantina Basin, Espinhaço supergroup– Ne/Brazil E. G. de Souza (2019)

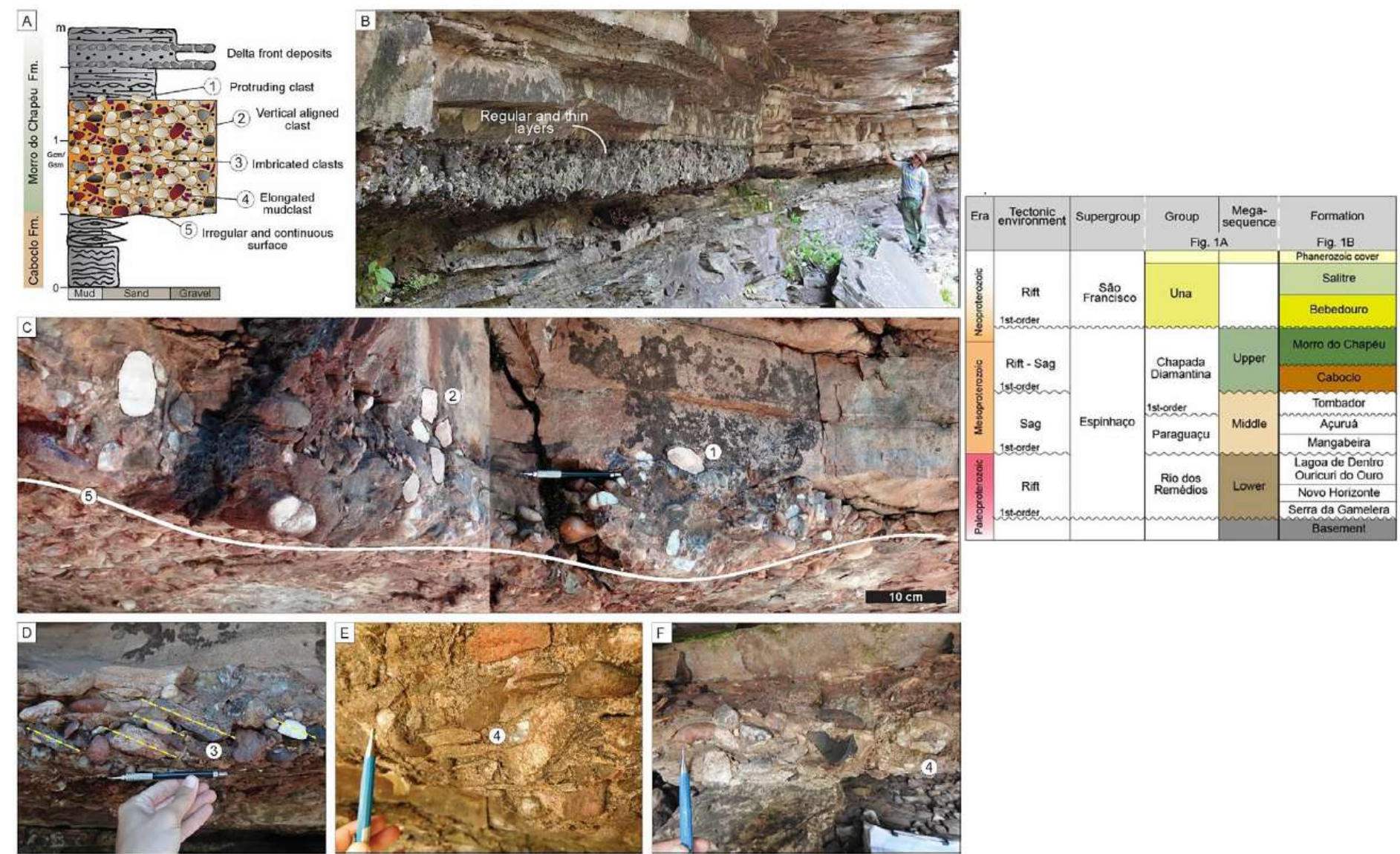


Fig. 4. A) Main characteristics of alluvial deposits (see Table 2); B) The single occurrence as a regular and thin layer with wide lateral extent, bounded by continuous and undulated surfaces; C) Layer details and representation of main larger clasts with vertically orientation; D) Imbricated and parallel clasts with the major axis to NW; E and F) Wide variation of form and composition of clasts, where the larger are subrounded to rounded while smaller are angular and low sphericity.



<https://geoportal.cprm.gov.br/geosgb/>

[https://geosgb.cprm.gov.br/geosgb/index\\_en.html](https://geosgb.cprm.gov.br/geosgb/index_en.html)



**THE END!**